

# Eicosanoic acid

<b>Other names:</b>	ARACHIC ACID Arachidic acid Arachidic acid (synthetic) Icosanoic acid N-EICOSANOIC ACID
<b>Inchi:</b>	InChI=1S/C20H40O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20(21)22/h2-19
<b>InchiKey:</b>	VKOBVWXKNCXXDE-UHFFFAOYSA-N
<b>Formula:</b>	C20H40O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCC(=O)O
<b>Mol. weight [g/mol]:</b>	312.53
<b>CAS:</b>	506-30-9

## Physical Properties

Property code	Value	Unit	Source
chl	-12574.20 ± 1.50	kJ/mol	NIST Webbook
gf	-148.22	kJ/mol	Joback Method
hf	-720.94	kJ/mol	Joback Method
hfus	53.24	kJ/mol	Joback Method
hvap	83.54	kJ/mol	Joback Method
log10ws	-7.29		Crippen Method
logp	7.113		Crippen Method
mcvol	300.100	ml/mol	McGowan Method
pc	1123.81	kPa	Joback Method
pt	2.00e-06 ± 1.07e-06	kPa	NIST Webbook
rinpol	377.80		NIST Webbook
rinpol	2365.30		NIST Webbook
rinpol	2380.00		NIST Webbook
rinpol	2365.30		NIST Webbook
rinpol	2359.00		NIST Webbook
rinpol	2359.00		NIST Webbook
rinpol	2380.00		NIST Webbook
rinpol	2366.00		NIST Webbook
rinpol	377.80		NIST Webbook
rinpol	2380.00		NIST Webbook
rinpol	2359.00		NIST Webbook
tb	803.05	K	Joback Method
tc	983.47	K	Joback Method

tf	348.40 ± 0.05	K	NIST Webbook
tt	347.25 ± 0.60	K	NIST Webbook
tt	348.23 ± 0.02	K	NIST Webbook
vc	1.181	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1037.42	J/mol×K	953.40	Joback Method
cpg	1007.04	J/mol×K	893.26	Joback Method
cpg	990.60	J/mol×K	863.19	Joback Method
cpg	973.29	J/mol×K	833.12	Joback Method
cpg	955.05	J/mol×K	803.05	Joback Method
cpg	1051.45	J/mol×K	983.47	Joback Method
cpg	1022.64	J/mol×K	923.33	Joback Method
cps	545.14	J/mol×K	298.15	NIST Webbook
dvisc	0.0004736	Paxs	488.77	Joback Method
dvisc	0.0001796	Paxs	551.62	Joback Method
dvisc	0.0000830	Paxs	614.48	Joback Method
dvisc	0.0000443	Paxs	677.34	Joback Method
dvisc	0.0000263	Paxs	740.19	Joback Method
dvisc	0.0016627	Paxs	425.91	Joback Method
dvisc	0.0000169	Paxs	803.05	Joback Method
hfust	69.20	kJ/mol	348.20	NIST Webbook
hfust	69.20	kJ/mol	348.20	NIST Webbook
hfust	71.60	kJ/mol	347.80	NIST Webbook
hsubt	200.00 ± 7.50	kJ/mol	336.80	NIST Webbook
hsubt	199.60 ± 7.50	kJ/mol	341.50	NIST Webbook
hsubt	148.40	kJ/mol	314.00	NIST Webbook
hvapt	125.50	kJ/mol	392.00	NIST Webbook
hvapt	114.50	kJ/mol	573.50	NIST Webbook
hvapt	143.70	kJ/mol	298.00	Vapor Pressures and Vaporization, Sublimation, and Fusion Enthalpies of Some Fatty Acids

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.85579e+01
Coeff. B	-7.38557e+03
Coeff. C	-1.34680e+02
Temperature range (K), min.	538.92
Temperature range (K), max.	692.23

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Vapor Pressures and Vaporization, Sublimation, and Fusion Enthalpies of some Fatty Acids:</b>	<a href="https://www.doi.org/10.1021/je300902c">https://www.doi.org/10.1021/je300902c</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=954">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=954</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C506309&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C506309&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pt:</b>	Triple Point Pressure

<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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